

synthesis of alkaloids is under investigation.

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Supplementary Material Available: Spectroscopic data for all new compounds (6 pages). Ordering information is given on any current masthead page.

Stereoselective Synthesis of Exocyclic Alkenes via Zirconium-Promoted Alkyl-Diene Coupling¹

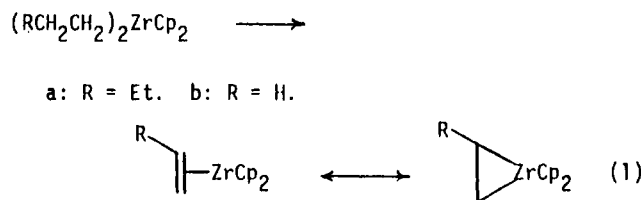
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Summary: The reaction of 1-vinyl-1-cycloalkenes with 1 equiv of $(RCH_2CH_2)_2ZrCp_2$, where R is H or alkyl, followed by treatment with electrophiles, e.g., proton donors, I_2 , and ketones, can provide the corresponding exocyclic alkenes in good yields, the stereoselectivity being $\geq 95\%$.

Sir: We have recently reported the reactions of "ZrCp₂", such as $n\text{-Bu}_2ZrCp_2$,^{2,3} with nonconjugated enynes,^{3,4} diynes,^{2,3,5} and dienes.¹ With $n\text{-Bu}_2ZrCp_2$ the actual "ZrCp₂" reagent has been shown to be $(\eta^2\text{-1-butene})ZrCp_2$ (1a), in which 1-butene serves as a ZrCp₂-protecting but "nonparticipating" ligand.^{3,6} On the other hand, the reaction of alkene-ZrCp₂ complexes (1) with monoalkenes incorporates the alkene moiety in the products.⁷



We now report that the reaction of 1 with conjugated dienes (2) gives zirconacycles represented by 3, which can be converted to a variety of organic products, such as 4-6 shown in Scheme I. Particularly noteworthy is that 1-vinyl-1-cycloalkenes can be readily converted to stereodefined exocyclic alkenes.^{8,9} Although zirconacycles represented by 3 have been previously prepared by the re-

Table I. Synthesis of Exocyclic Alkenes by the Reaction of Conjugated Dienes with Dialkylzirconocenes

conjugated diene	R of R_2ZrCp_2	exocyclic alkene	yield, %	
			GLC	isolated
2a	Et	4a	91	56
2a	Et	7	-	55
2a	Et	8	-	66
2a	Et	5a	55	48
2a	<i>n</i> -Bu	4b	90	80
2b	Et	4c	82	67
2b	<i>n</i> -Bu	4d	89	72
2c	Et	4e	67	64
2c	<i>n</i> -Bu	4f	-	70
2d	<i>n</i> -Bu	4g	83	68

action of alkenes with conjugated diene-ZrCp₂ complexes obtained by either the enediylmagnesium-Cl₂ZrCp₂ reaction^{10a} or the diene-Ar₂ZrCp₂ reaction,^{10b} the conversion of 2 to 3 as shown in Scheme I represents a novel route. Furthermore, the conversion of 1-vinyl-1-cycloalkenes to stereodefined exocyclic alkenes has not been previously reported.

Typically, ethyllithium in Et₂O (1.0 M, 6.0 mL, 6.0 mmol) was added to a solution of Cp₂ZrCl₂ (0.88 g, 3.0 mmol) in 10 mL of THF at -78 °C. After having been stirred for 1 h at -78 °C, the mixture was treated with 1-ethenyl-1-cyclohexene (0.325 g, 3.0 mmol), and warmed to 25 °C over a few hours. Examination of the reaction mixture by ¹H NMR spectroscopy using benzene as an internal standard indicated the formation of a ZrCp₂ derivative exhibiting two singlets for the Cp groups at δ 5.50 and 5.65 in >90% yield. The compound has been tentatively identified as 3a on the basis of the following. The reaction mixture obtained above was quenched with a mixture of 2 N HCl and Et₂O, extracted with Et₂O, washed with aqueous NaHCO₃ and brine, and dried over MgSO₄. Distillative workup gave 0.23 g (56%) of butylidene-cyclohexane: IR (neat) 3030 (w), 1675 (w), 1450 (s), 1370 (m) cm⁻¹; ¹H NMR (CDCl₃, Me₄Si) δ 0.88 (t, J = 7 Hz, 3 H), 1.25-1.4 (m, 2 H), 1.4-1.7 (m, 6 H), 1.9-2.0 (m, 2 H), 2.0-2.2 (m, 4 H), 5.07 (t, J = 7 Hz, 1 H); ¹³C NMR (CDCl₃, Me₄Si) δ 13.75, 27.89, 28.11, 28.17, 28.77, 29.18, 37.24, 121.27, 139.56. Treatment of 3a with D₂O (1 mL/3 mmol of diene, 25 °C, 2 H) followed by workup with 2 N HCl produced a >95% pure monodeuterio derivative of 4a, i.e., 7, in 55% isolated yield. The use of 2 N DCl in place of HCl cleanly produced the dideuterio derivative 8 in 66% yield, the extents of deuterium incorporation on the ring

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- (5) For a related study by other workers, see Nugent, W. A.; Thorn, D. L.; Harlow, R. L. *J. Am. Chem. Soc.* 1987, 109, 2788.
- (6) 1-Butene is not incorporated in the product and recovered as such. (7) Swanson, D. R.; Rousset, C. J.; Negishi, E.; Takahashi, T.; Seki, T.; Saburi, M.; Uchida, Y. *J. Org. Chem.* 1989, 54, 3521.
- (8) For a review, see Negishi, E. In *Advances in Metal-Organic Chemistry*; Liebeskind, L. S., Ed.; JAI Press: Greenwich, CT, 1988; Vol. 1, p 177.
- (9) For our previous papers on the synthesis of exocyclic alkenes other than refs 2-4, see: (a) Luo, F. T.; Negishi, E. *J. Org. Chem.* 1983, 48, 5144. (b) Miller, J. A.; Negishi, E. *Isr. J. Chem.* 1984, 24, 76. (c) Negishi, E.; Zhang, Y.; Cederbaum, F. E.; Webb, M. B. *J. Org. Chem.* 1986, 51, 4080. (d) Negishi, E.; Zhang, Y.; Bagheri, V. *Tetrahedron Lett.* 1987, 28, 5793. (e) O'Connor, B.; Zhang, Y.; Negishi, E. *Tetrahedron Lett.* 1988, 29, 3903. (f) Zhang, Y.; Negishi, E. *J. Am. Chem. Soc.* 1989, 111, 3454. (g) Zhang, Y.; Miller, J. A.; Negishi, E. *J. Org. Chem.* 1989, 54, 2043.

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Scheme I

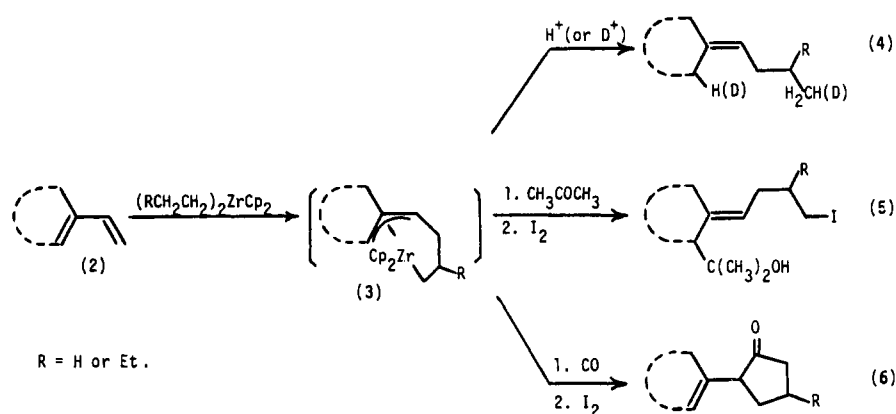
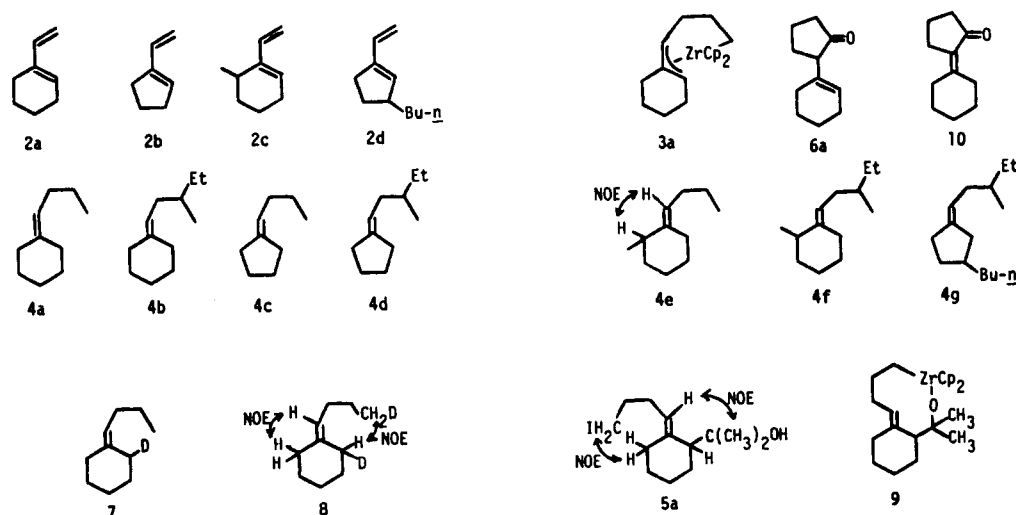


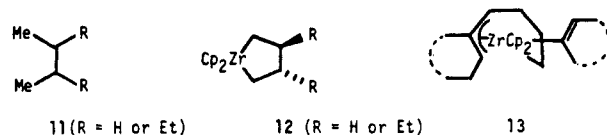
Chart I



and the chain terminus were both $\geq 95\%$. Interestingly, treatment of **3a** first with dry acetone and then with I_2 provided a 55% yield of **5a**, in which the sense of alkene geometry is opposite to that in **4a**. The stereochemistry of **5a** corresponds to the probable intermediacy of **9**. The reaction of **3a** with CO (1.1 atm) at 0°C for 2 h followed by iodinolysis (1.1 equiv) gave a 47% yield of **6a** and its regioisomer **10** in a 90:10 ratio (see Chart I).

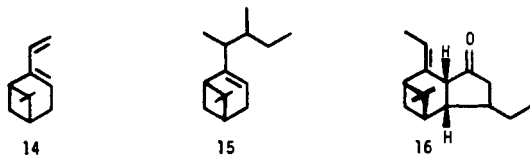
The results of the conversion of conjugated dienes into exocyclic alkenes **4a-g** as well as **5a**, **7**, and **8** are summarized in Table I. The most noteworthy feature is that all of the exocyclic alkene products are $\geq 95\%$ isomerically pure single isomers, as judged by analysis of ^1H and ^{13}C NMR spectra as well as GLC. The results indicate that the reaction is not only $\geq 95\%$ regioselective but also $\geq 95\%$ stereoselective in cases where the exocyclic alkenes are stereodefined, as in the cases of **4e-g**, **5a**, **7**, and **8**. The stereochemical assignment is based on ^1H 2D NOESY NMR analysis of selected products, i.e., **4e**, **5a**, and **8**. Highly intriguing is that even **4f** and **4g** exhibit only one set each of ^{13}C NMR signals. Although the relative stereochemical relationship between two asymmetric carbon centers has not been established, these compounds must be single diastereomers. The required regiodefined dienes were prepared by the Pd-catalyzed reaction of vinylzinc chloride^{11a} with the corresponding cycloalkenyl triflates.^{11b} That the conversion of conjugated dienes into alkylated

alkenes is not limited to cyclic derivatives is indicated by the preparation of 2,5-dimethyl-2-heptene in 83% GLC yield (53% isolated) via the *n*-Bu₂ZrCp₂-isoprene reaction. The "pair" selectivity of the reaction is also $\geq 95\%$. Only traces of the homocoupled products, i.e., **11**, and dimers of **2** were formed. The observed high "pair" selectivity is readily explained on the assumptions that **4**, which must be allylically stabilized, is more stable than **12** and that the 1:1 reactant ratio would dictate the amount of the other possible pair, i.e., **13**, to be the same as that of **12**. The high "pair" selectivity is reminiscent of the *n*-Bu₂ZrCp₂-styrene reaction reported recently by us,⁷ in which benzylic stabilization is thought to play a similar role.



Only in the reaction of **14** were some abnormal results observed. Thus, its reaction with *n*-Bu₂ZrCp₂ at -78 to 0°C followed by protonolysis produced **15** in 81% GLC yield (46% isolated) which was isomerically $\geq 95\%$ pure. Even more puzzling is that the cyclization reaction followed by carbonylation produced isomerically $\geq 95\%$ pure **16** as essentially the only bicyclic carbonylation product, albeit only in 27% isolated yield. Its identification is primarily based on ^1H and ^{13}C NMR analysis including extensive decoupling and 2D NOESY experiments. At present, we cannot readily rationalize its formation.

(11) (a) For a review of Pd-catalyzed cross coupling, see: Negishi, E. *Acc. Chem. Res.* **1982**, *15*, 340. (b) McMurry, J. E.; Scott, W. J. *Tetrahedron Lett.* **1983**, *24*, 979.



The present study adds to the growing list of highly stereoselective methods for preparing exocyclic alkenes.⁹ Finally, we made an erroneous claim that the reaction of 1,3-butadiene with *n*-Bu₂ZrCp₂ gave (1,3-butadiene)ZrCp₂.²

In the light of the results presented herein, we wish to correct this error with apology.

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Total Synthesis of 13-Oxygenated Prostanoids Derived from Arachidonate: An Instance of Extraordinary Variability in the Stereochemical Sense of a Mukaiyama Aldol Reaction

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Summary: A complete reversal of stereochemical outcome in the Mukaiyama reaction of oct-2-ynal as opposed to either (*E*)- or (*Z*)-oct-2-enal with a common enoxysilane has been noted and applied to a very straightforward synthesis of the titled series.

Sir: The principal 9 α ,11 α -endoperoxides arising from the in vivo oxidation of arachidonate, in the presence of PGH synthase, contain a trans 13,14-double bond and either 15(*S*)-hydroperoxy (PGG₂) or 15(*S*)-hydroxy (PGH₂) functions.^{1,2} Surprisingly, it was recently shown that this process also produces isomers of the above, bearing oxygen substitution at C₁₃ (Scheme I). Reduction of this endoperoxide gives rise to the allylic isomer of PGF_{2 α} .^{2,3} That this prostaglandin is properly represented by structure **1** was demonstrated by Hofmann and colleagues. The four permutants bearing a 13-hydroxy group and a Δ^{14} double bond were synthesized.⁴ A minor synthetic product, shown to be compound **1**, corresponded to the naturally derived product.

It was our intention to provide the difficultly accessible compound **1** through total synthesis. Recently we described two concise syntheses of PGF_{2 α} ⁵ (Scheme II). Each synthesis involved a stereospecific aldol-like reaction,⁶ catalyzed by titanium tetrachloride, between enoxysilane **4**⁷ and the α,β -unsaturated aldehydes (*Z*)- and (*E*)-2-

octenal. Each aldol reaction occurred with transfer of the triethylsilyl group to afford **5** and **6**, respectively, wherein, in each case, the configuration at C₁₃ is *R*.⁸ In principle, one could envision one of several protocols to bring about an overall inversion at C₁₃ with preservation of the *Z*- Δ^{14} unsaturation for the conversion of compound **4** to **1**. In practice a much simpler route was discovered.

The key reaction was that of enoxysilane **4** with oct-2-ynal (**7**).⁹ As in the previous work,⁵ the reaction was carried out in methylene chloride in the presence of titanium tetrachloride (1 equiv). There was thus obtained an acetylenic alcohol. Unlike the reactions with the two enals, we could not observe any of the silyl group transfer product. At this stage, it was not possible to determine the stereochemistry at C₁₃ in this product. Semihydrogenation of the triple bond (H₂; Lindlar catalyst; 50 min) afforded a *Z*-allylic alcohol, which, upon acetylation (Ac₂O, Py, DMAP), afforded a *Z*-allylic acetate in 50% overall yield from **3**. This compound did not converge with any transformation products of **5**. Our suspicion that the noncorrespondence arose from a differing configuration at C₁₃⁸ was confirmed. The allylic acetate, thus formulated as **9**, upon treatment with PdCl₂(MeCN)₂¹⁰ afforded compound **11**. The same compound was obtained from the allylic transposition carried out in the same way on the 13-acetate derived from **6**.⁵ The structure of **11** is secure in that it had been converted to PGF_{2 α} ⁵ by a sequence that

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(7) As shown, this compound is prepared from a group transfer Michael reaction of a suitable silylketene acetal. Cf. Webster, O. W.; Hertler, W. R.; Sogah, D. Y.; Farnham, W. B.; Rajan Babu. *J. Am. Chem. Soc.* 1983, 105, 5706.

(8) The reader will not be confused at the fact that the *R* and *S* descriptors for compounds **6**, **7**, and **8** (13*R* in each case) fail to communicate the fact that the relative configuration at C₁₃ is opposite in **8** from that of **5** and **6**. This situation is better captured by focusing on the C₁₂-C₁₃ relationship, which may be said to be *syn* in **5** and **6** (in the zig-zag conformer shown) and *anti* in **8**.

(9) Prepared by acid-catalyzed hydrolysis of the corresponding dimethyl acetal. Bryne, B.; Lafleur Lawter, L. M.; Wengenroth, K. *J. Org. Chem.* 1986, 51, 2607.

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